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To cite this article: Mert Olgun Karataş, Harun Uslu, Bülent Alıcı, Başak Gökçe, Nahit Gencer & Oktay Arslan (2016) Some coumarins and benzoxazinones as potent paraoxonase 1 inhibitors, Journal of Enzyme Inhibition and Medicinal Chemistry, 31:6, 1386-1391, DOI: [10.3109/14756366.2016.1142982](https://doi.org/10.3109/14756366.2016.1142982)

To link to this article: <https://doi.org/10.3109/14756366.2016.1142982>



Published online: 17 Feb 2016.



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RESEARCH ARTICLE

## Some coumarins and benzoxazinones as potent paraoxonase 1 inhibitors

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### Abstract

In this study, we aimed to investigate the effect of some coumarin and benzoxazinone derivatives on the activity of human PON1. Human serum paraoxonase 1 was purified from fresh human serum blood by two-step procedures that are ammonium sulfate precipitation (60–80%) and then hydrophobic interaction chromatography (Sephacrose 4B, L-tyrosine and 1-naphthylamine). The enzyme was purified 232-fold with a final specific activity of 27.1 U/mg. *In vitro* effects of some previously synthesized ionic coumarin or benzoxazinone derivatives (**1–21**) on purified PON1 activity were investigated. Compound **14** (1-(2,3,4,5,6)-pentamethylbenzyl-3-(6,8-dimethyl-2H-chromen-2-one-4-yl))benzimidazolium chloride was found out as the strongest inhibitor ( $IC_{50} = 7.84 \mu M$ ) for PON1 among the compounds. Kinetic investigation and molecular docking study were evaluated for one of the most active compounds (compound **12**) and obtained data showed that this compound is competitive inhibitor of PON1 and interact with Leu262 and Ser263 in the active site of PON1. Moreover, coumarin derivatives were found out as the more potent inhibitors for PON1 than benzoxazinone derivatives.

### Keywords

Benzoxazinone, coumarin, docking, inhibition, paraoxonase

### History

Received 7 September 2015  
Revised 6 January 2016  
Accepted 7 January 2016  
Published online 11 February 2016

### Introduction

The paraoxonase (PON) enzyme family has three members, PON1, PON2 and PON3, that PON1 is the best studied member of this family within mammalian enzymes. The PON1 is the calcium-dependent enzyme that has 355 amino acids with a mass of 43 kDa<sup>1,2</sup>. ‘‘PON’’ gets name from one of the most commonly used *in vitro* substrates, paraoxon<sup>3</sup>. It also hydrolyzes aromatic carboxyl esters such, lactones, thiolactones and cyclic carbonate esters<sup>4–7</sup>. Furthermore, PON1 hydrolyzes some organophosphate-derived nerve agents such as sarin and soman and pesticide diazoxon<sup>1</sup>; however, physiological substrate and biological function of this enzyme have not known yet. Beside hydrolysis of organophosphates, antiatherogenic properties of PON1 were reported. Some *in vitro* studies showed that PON1 inhibits oxidation of low-density lipoprotein (LDL) and high-density lipoprotein (HDL) and reduced levels of oxidized lipids are involved in the initiation of atherosclerosis<sup>8,9</sup>. Moreover, decreased PON1 levels and activities were reported on the atherosclerotic patients<sup>10</sup>.

In light of these knowledge, decreased activity of PON1 was acknowledged as a risk of atherosclerosis and organophosphate toxicity, and therefore, the factors affecting PON1 have to be understood clearly. In literature, some crucial studies were carried out in order to understand the biological function and substrate, mechanisms of action of PON1<sup>11,12</sup>. A small number of inhibition

studies on PON1 were reported in literature with various commercially available drugs, some metal ions, coumarin derivatives and pesticides<sup>13–18</sup>. Because of all these reasons, PONs have become the subject of intensive research area.

Coumarins are member of a class of compounds known as benzopyrones. In recent year’s research, the coumarin chemistry have attracted much attention due to diverse biological and pharmacological properties of coumarin derivatives, such as anticoagulant<sup>19</sup>, anticancer<sup>20–22</sup>, anti-HIV<sup>23,24</sup>, anti-inflammatory<sup>25</sup>, antimicrobial<sup>26,27</sup> and some different biological activities<sup>28–32</sup>. Most recently, coumarins were reported as ‘‘suicide’’ inhibitors of human carbonic anhydrases (hCA)<sup>33</sup>. As shown, coumarin derivatives are promising candidates for the treatment of some important disease.

The other class of compounds, which was used in this work, is benzoxazinone. This compound is a heterocyclic compound that contains the 1,4-oxazinone ring fused to benzene. Various biological properties of benzoxazinone derivatives are known in the literature and they have an important role in the design and development of new drugs<sup>34–38</sup>.

As mentioned previously, the activity of PON1 is vital for human and we aimed to examine the *in vitro* effects of 21 coumarin and benzoxazinone derivatives on the activity of purified human serum PON1. We chose these class compounds because of their important biological activities. Moreover, in our previous study, we showed that some coumarin compounds have significant inhibitory properties on the activity of PON1<sup>39</sup>. Another reason of carrying out this study is biological interference of coumarin compounds with coagulation, thrombotic events

and antiatherogenic properties of PON1. The compounds that were used in this study had remained from our previous study<sup>40</sup>. For one of the strongest inhibitors, we performed kinetic and molecular docking studies to obtain insights into binding mode of these compounds with key residues of the active site of PON1.

## Materials and methods

The materials used include Sepharose 4B, *L*-tyrosine, 1-naphthylamine, paraoxon and protein assay reagents were obtained from Sigma Chem. Co (Istanbul, Turkey). Thirteen coumarin and eight benzoxazinone derivatives were ready from our previous study and detailed<sup>40</sup>.

### Paraoxonase enzyme assay

Paraoxonase enzyme activity toward paraoxon was quantified spectrophotometrically by the method described by Gan et al.<sup>41</sup>. The enzyme assay was based on the estimating of *p*-nitrophenol at 412 nm. The molar extinction coefficient of *p*-nitrophenol ( $\epsilon = 17100 \text{ M}^{-1} \text{ cm}^{-1}$  at pH 8) was used to calculate enzyme activity. The reaction was followed for 2 min at 37 °C by monitoring the appearance of *p*-nitrophenol at 412 nm in automated recording spectrophotometer (Biotek, Winooski, VT). About 2 mM of final substrate concentration was used during enzyme assay, and all measurements were taken in duplicate and corrected for the nonenzymatic hydrolysis.

### Purification of paraoxonase from human serum by hydrophobic interaction chromatography

Human serum was isolated from 40 mL fresh human blood and put into a dry tube. The blood samples were centrifuged at 3000 rpm for 15 min and the serum was removed. Firstly, serum paraoxonase was isolated by ammonium sulfate precipitation (60–80%). The precipitate was collected by centrifugation at 15 000 rpm for 20 min and redissolved in 100 mM Tris HCl buffer (pH 8.0). Then, we synthesized the hydrophobic gel, including Sepharose 4B, *L*-tyrosine and 1-naphthylamine, for the purification of human serum paraoxonase<sup>42</sup>. The purified enzyme had a specific activity of 11.76 U/mg. The column was equilibrated with 0.1 M of a  $\text{Na}_2\text{HPO}_4$  buffer (pH 8.00) including 1 M ammonium sulfate. The paraoxonase was eluted with an ammonium sulfate gradient using 0.1 M  $\text{Na}_2\text{HPO}_4$  buffer with and without ammonium sulfate (pH 8.00). The purified PON1 enzyme was stored in the presence of 2 mM calcium chloride in order to maintain activity.

### Total protein determination

The absorbance at 280 nm was used to monitor the protein in the column effluents and ammonium sulfate precipitation. Quantitative protein determination was achieved by absorbance measurements at 595 nm according to Bradford<sup>43</sup>, with bovine serum albumin as a standard.

### SDS polyacrylamide gel electrophoresis

SDS polyacrylamide gel electrophoresis was performed after purification of the enzyme. It was carried out in 10% and 3% acrylamide concentration for the running and stacking gel, respectively, containing 0.1% SDS according to Laemmli<sup>44</sup>. A 20 mg sample was applied to the electrophoresis medium. Gel was stained overnight in 0.1% Coomassie Brilliant Blue R-250 in 50% methanol and 10% acetic acid, then destained by frequently changing the same solvent, without dye. The electrophoretic pattern was photographed with the system of produce as an image of the gel.

### *In vitro* kinetic studies and calculation of IC<sub>50</sub> and *K<sub>i</sub>* values

For the inhibition studies of coumarin derivatives, different concentrations of coumarin derivatives were added to the reaction medium. PON1 activity with coumarin derivatives was assayed by following the hydration of paraoxon. Activity percentage values of PON for five different concentrations of each coumarin derivatives were determined by regression analysis using the Microsoft Office 2000 Excel. PON1 enzyme activity without a coumarin derivative was considered as 100% activity. The inhibitor concentration causing up to 50% inhibition (IC<sub>50</sub> values) for coumarin derivatives were determined from the graphs. So as to determinate *K<sub>i</sub>*, *K<sub>m</sub>* and *V<sub>max</sub>* values of the PON1 enzyme using paraoxon as a substrate was measured at eight different substrate concentrations (0.5, 0.6, 0.7, 0.8, 0.9, 1.0, 1.1 and 1.2 mM) at pH 8.00 and 37 °C in the presence of two different coumarin concentrations of compound **12** (28.5 μM and 19 μM) as knowing highest and lowest final concentrations. *K<sub>m</sub>* and *V<sub>max</sub>* values were calculated 62.07 μM, 6.66 μmol/min, respectively, through Lineweaver–Burke graphs. The coumarin compound named **12** inhibited PON1 enzyme activity in a competitive inhibition type with *K<sub>i</sub>* of 13.72 μM value. As a result of *in vitro* experiments and by the means of Lineweaver–Burke graphs competitive inhibition type was controlled with docking for confirmation.

### Explanation of using compound **12** instead of compound **14** for determination of *K<sub>i</sub>* value

Although there were more effective inhibitors, compound **12** was chosen for estimation *K<sub>i</sub>*, *K<sub>m</sub>* and *V<sub>max</sub>* values. The only reason was run short of these inhibitors. So that compound **12** was used for the most potent inhibitor in depleted compounds after inhibition studies (estimation for IC<sub>50</sub> values).

### Molecular docking

Enzyme setup. Determination of the consistent receptor was based on previous study<sup>12</sup>. Macromolecule file (PDB code: 1V04) was modified using the ADT package version 1.5.6rc3 (Ankara, Turkey). All water molecules were deleted and polar hydrogens were added. Subsequently, Gasteiger charges were calculated and the generated pdbqt files were saved.

Ligand Energy minimization of compound **12** was carried out using GAMESS module for ChemOffice version Ultra 8.0.3 (Ankara, Turkey). All data were saved as pdb with the aid of Molegro Molecular Viewer version 2.5 (Ankara, Turkey). Further modification of these partial charges of pdb files was carried out through the ADT package so that the charges of the nonpolar hydrogen atoms allocated to the atom to which the hydrogen is attached. These modified pdb files saved as pdbqt files. Appropriate grid box points were determined by centering on ligand separately for each compound.

## Results and discussion

As mentioned in ‘Introduction’ section, a few *in vitro* inhibition studies on PON1 were reported in literature so far. In a study, Erzen and coworkers studied inhibition effects of some coumarin derivatives and 6,7-dihydroxy-3-(4-methylphenyl)-2H-chromen-2-one inhibited PON1 with IC<sub>50</sub> value of 3 μM<sup>14</sup>. According to our literature survey, this compound is the most active inhibitor of human PON1. In addition, inhibition effects of some anesthetics, calcium channel blockers, some metal ions and some pesticides on PON1 were reported and IC<sub>50</sub> values of 85–1678, 121–255, 4–317 and 709–35011 μM were determined, respectively<sup>15–18</sup>.

Table 1. Structures and IC<sub>50</sub> values of compounds. These compounds were ready from our previous study<sup>35</sup>.

Compound	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	A	IC <sub>50</sub> (µM)
<b>1</b>	-CH <sub>3</sub>	-H	-	dmc	21.51
<b>2</b>	-CH <sub>3</sub>	-CH <sub>3</sub>	-	dmc	238.61
<b>3</b>	-CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	-H	-	dmc	11.74
<b>4</b>	-CH <sub>3</sub>	-H	-	dhc	11.00
<b>5</b>	-CH <sub>3</sub>	-CH <sub>3</sub>	-	dhc	293.40
<b>6</b>	-CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	-H	-	dhc	19.09
<b>7</b>	-CH <sub>3</sub>	-H	-	box	218.13
<b>8</b>	-CH <sub>3</sub>	-CH <sub>3</sub>	-	box	307.98
<b>9</b>	-CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	-H	-	box	347.31
<b>10</b>	-	-	-CH <sub>3</sub>	dmc	16.54
<b>11</b>	-	-	-CH <sub>2</sub> CH=CH <sub>2</sub>	dmc	16.27
<b>12</b>	-	-	-C <sub>6</sub> H <sub>5</sub>	dmc	12.83
<b>13</b>	-	-	-C <sub>10</sub> H <sub>7</sub> -2-yl	dmc	12.59
<b>14</b>	-	-	-C <sub>6</sub> (CH <sub>3</sub> ) <sub>5</sub> -2,3,4,5,6	hc	7.84
<b>15</b>	-	-	-C <sub>10</sub> H <sub>7</sub> -2-yl	hc	30.02
<b>16</b>	-	-	-C <sub>10</sub> H <sub>7</sub> -2-yl	dhc	17.50
<b>17</b>	-	-	-CH <sub>3</sub>	box	240.55
<b>18</b>	-	-	-CH <sub>2</sub> CH=CH <sub>2</sub>	box	543.70
<b>19</b>	-	-	-C <sub>6</sub> H <sub>5</sub>	box	466.18
<b>20</b>	-	-	-C <sub>10</sub> H <sub>7</sub> -2-yl	box	26.31
<b>21</b>	-	-	-C <sub>6</sub> H <sub>2</sub> (OCH <sub>3</sub> ) <sub>3</sub> -3,4,5	box	36.57

In this study, the effects of 21 ionic coumarin or benzoxazinone derivatives on PON1 were investigated. These compounds were ready from our previous study. The compounds had been synthesized by the interaction of *N*-alkyl imidazolium or benzimidazolium derivatives with some 4-chloromethylene substituted coumarin derivatives or 6-chloroacetylene substituted benzoxazinone derivatives. The synthesized compounds had been obtained in good yields between 43–92%. Characterization of the compounds had been elucidated by nuclear magnetic resonance (NMR), infrared (IR) and elemental analyses in satisfactory manner<sup>40</sup>. For evaluating the PON1 activity, all compounds were subjected to PON1 inhibition assay with paraoxon as substrate. The IC<sub>50</sub> values and structures of compounds were given in Table 1. The results showed that all compounds (**1–21**) inhibited PON1 enzyme activity with different sensitivity. We determined IC<sub>50</sub> values in the range of 7.84–543.70 µM. Among them, 13 compounds (**1,3,4,6,10–16,20,21**) performed high inhibition activity and IC<sub>50</sub> values of these compounds are ranging between 7.84–36.57 µM. Compound **14** was found out as the strongest inhibitor of PON1 with 7.84 µM value of IC<sub>50</sub>. The other eight compounds showed relatively lower inhibitory activity against PON1. Moreover, *K<sub>i</sub>* value and inhibition type of compound **12** (1-benzyl-3-(6,8-dimethyl-2H-chromene-2-one-4-yl)benzimidazolium chloride were investigated and results demonstrated that this compound inhibited enzyme activity in a competitive manner with *K<sub>i</sub>* value of 13.72 µM (Figure 1). As mentioned earlier,

inhibitory activities of some biologically active compounds and metal ions on PON1 were studied but, studies about interactions of previously unused compounds with PON1 are scarce. In literature, it is known that coumarin derivatives have high inhibitory properties against PON1 but, mechanism of action is still unknown. We may say that, the compounds which reported in this study have good inhibitory properties when results compared with literature.

In a study, it was shown that simple lactone derivatives were hydrolyzed by PON1. In the same study, authors reported that lactam derivatives and a coumarin were not hydrolyzed by PON1<sup>11</sup>. In our study, tested compounds are bearing several types of substituent, such as hydroxyl, azolium and alkyl or aryl groups, so according to the results of our and previous studies, compounds **1–21** are not suitable substrates for PON1 and some of them inhibited PON1 effectively.

### Structure–activity relationship

We can classify the compounds tested in this study in two different ways; (i) imidazolium- or benzimidazolium-based compounds, (ii) coumarin- or benzoxazinone-based compounds. From results of enzyme inhibition assay, two opinions about structure–activity relationship were emerged.

In the tested compounds, some compounds are bearing same groups in their structures apart from coumarin or benzoxazinone

Figure 1. Inhibition of paraoxonase by compound **12**. The slope of Lineweaver–Burk plots indicates competitive inhibition for paraoxon.

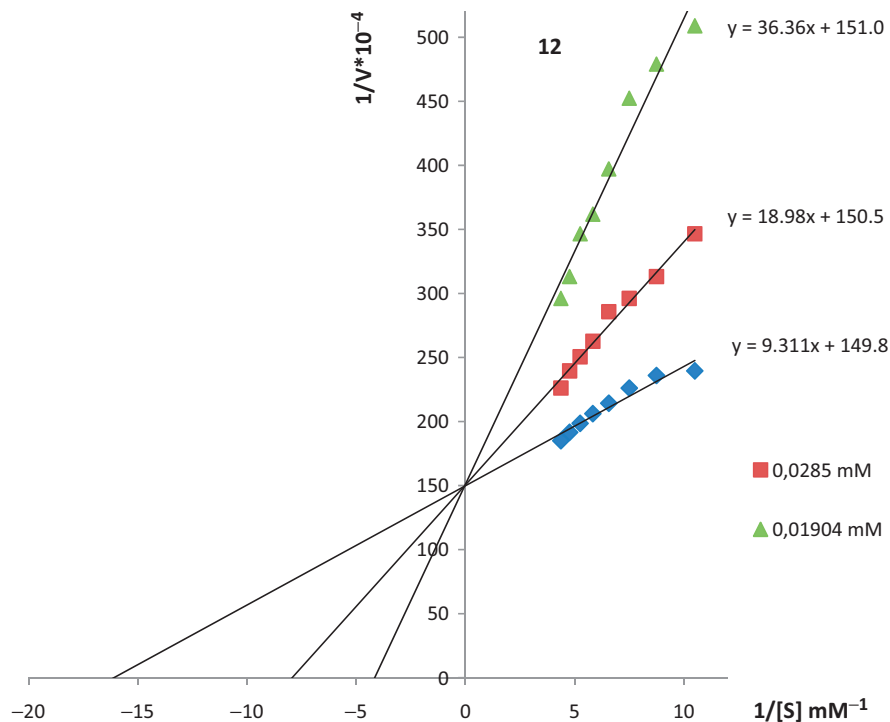


Table 2. Kinetic and molecular docking data of compound **12**.

Compound	K <sub>i</sub> (μM)	Inhibition type	Docking score (kcal/mole)	Close van der Waals contacts
<b>12</b>	13.72	Competitive	-4.7	Leu262, Ser263

scaffolds and we can use these compounds for comparison. When we compared the compounds **10–13** with **17–20**, each of these compounds are bearing methyl, allyl, benzyl and naphthyl groups as substituent and 6,8-dimethyl coumarin derivatives (**10–13**) have much lower IC<sub>50</sub> values than benzoxazinone derivatives (**17–20**). In addition, this situation can be seen in the imidazolium salts. When compounds **1–6** compared with compounds **7–9**, it is clear that coumarin-based imidazolium salts are more active than benzoxazinone corresponding. Therefore, it must be noted that coumarin derivatives showed higher inhibitory activity than benzoxazinone derivatives. If we compare coumarin benzimidazolium salts, there are some minor differences due to electronic effects. If compounds **15** and **16** compared, dihydroxy-substituted **16** is more active than monohydroxy-substituted **15**. Moreover, when compounds **14** and **15** compared which are monohydroxy coumarin substituted, pentamethyl substituted **14** is more active than naphthalene substituted **15**. If we want to observe the difference between imidazole and benzimidazole scaffold, we may compare the compounds **1** and **10**. These compounds are bearing same groups on nitrogen atom. IC<sub>50</sub> values showed that benzimidazole derivative **10** is more active than imidazole derivative **1**.

When imidazolium-based compounds **2,5** compared with **1,3,4,6**, an interesting difference can be seen. Compounds **2,5** performed lower activity than other imidazolium coumarin derivatives. In these compounds, methylation of position-2 decreased inhibition activity and possible explanation for the lower activity associated with the absence of acidic imidazolium proton. This fragment of compounds is the most hydrophilic and polar part of compounds and absence of acidic proton may cause reduced interactions between inhibitors and amino acid residues.

### Molecular docking studies

Kinetic data demonstrated that compound **12** is competitive inhibitor of PON1 and in order to obtain more insights into the binding mode, molecular docking studies were also performed for compound **12**. Docking scores were obtained using Lamarckian Genetic Algorithm and scoring function of AutoDock 4. Then, interactions were checked with the aid of ADT and Discovery Studio 4.0 Client (Ankara, Turkey). Docking scores and binding interactions of compound **12** with PON1 (PDB code: 1V04) are presented in Table 2. Final images of compound **12** for binding interactions are shown in Figure 2.

As seen from the data presented in Table 2, there are two amino acid residues participating in close van der Waals (<4Å) contacts with the inhibitor compound **12** when bound to the active site of PON1. According to results of molecular docking analysis, compound **12** interacts with Leu262 and Ser263 in the active site of PON1.

In 2004, Harel et al reported that the crystal structure of recombinant PON1 (rePON1) obtained from *Escherichia coli* (*E. coli*). In the same study, the authors reported that rePON1 and human PON1 (hPON1) have some differences in the sequence of amino acids, but they also reported that sequence variations between rePON1 and hPON1 are in the areas where their active site and overall structures are not affected. One of these differences is in the residue 263. It was shown that in the crystal structure of rePON1 residue 263 is serine and in hPON1 aspartic acid takes place of serine<sup>12</sup>. Although molecular docking studies report the interaction of compound **12** with serine, in hPON1, most probably it interacts with aspartic acid. Aspartic acid has a carboxyl group and this amino acid is negatively charged at pH 7.4 and this may cause more effective interaction with positively

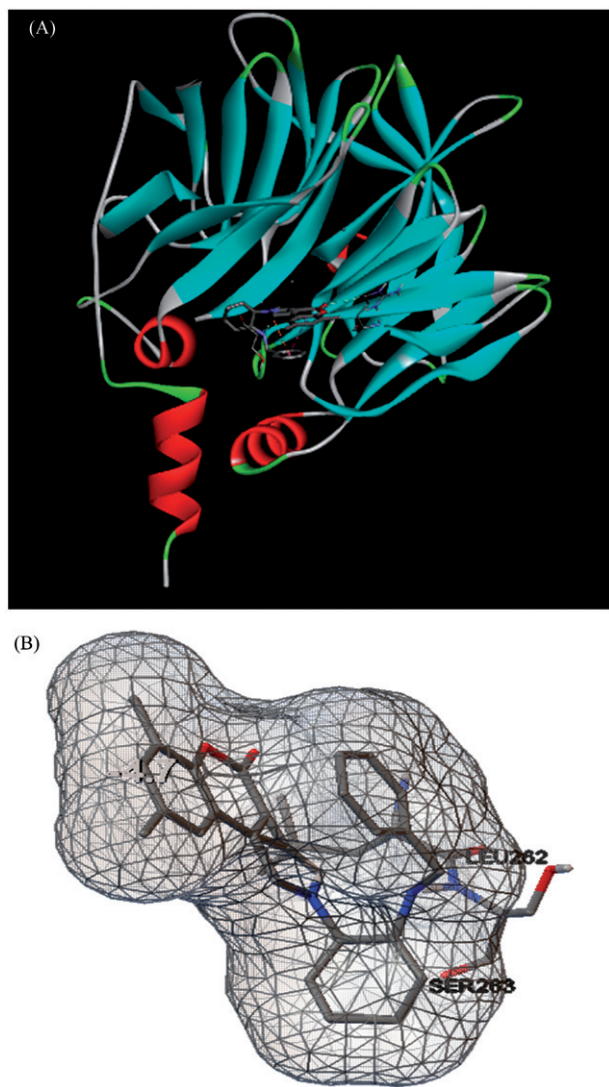


Figure 2. Docking of compound **12** within the active site of PON1 (A) Discovery Studio 4.0 Client images and (B) ADT images.

charged compound **12**. Moreover, decrease in the inhibition activity of compounds **2** and **5**, which were pointed out above may attributable with the absence of interaction of acidic proton with negatively charged aspartic acid. In addition, results showed that compound **12** did not interact with  $\text{Ca}^{2+}$  which is responsible for catalytic activity.

## Conclusion

PON1 has important detoxification role in metabolism and it is thought that decreased activity of PON1 may cause atherosclerosis and organophosphate toxicity so all factors affecting PON1 have to be well determined. Enzyme inhibition is an important issue for drug design and biochemical applications<sup>45–47</sup>. In this study, we have showed that human PON1 was inhibited effectively by some coumarin or benzoxazinone derivatives. Coumarin derivative compound **12** has been showed as competitive inhibitor of PON1. Molecular docking data showed that compound **12** inhibits PON1 by not-interacting with active site metal  $\text{Ca}^{2+}$  ions and it interacts with Leu262 and Ser263 in the active site of PON1. It is known that coumarins are toxic for humans in certain levels, but their some important biological activities make them attractive for use in future. Inhibition of PON1 by coumarin derivatives provides a justification to further consideration of limitation dosage of coumarin as drug and as flavor cause of risk

assessment, but more detailed investigations is necessary about mechanisms of inhibition of PON1 by coumarin and benzoxazinone derivatives.

## Declaration of interest

This work was supported by the Research Fund of the Balikesir University (project no. 2014–54).

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